

Low-Energy Solvents for CO₂ Capture Enabled by a Combination of Enzymes and Ultrasonics

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NETL CO₂ Capture
Technology Meeting
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Notices

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Agenda



- Project Overview
- Technology Background
- Progress and Status
 - Schedule and summary
 - Initial techno-economic assessment
 - Supporting lab results
 - Ultrasonic system testing at PNNL
 - University of Kentucky bench-scale system
- Conclusions & Next Steps

Novozymes in Brief – World Leader in Bioinnovation

Producing large volume enzymes for industrial applications

1. Improving the production host

Improving the microorganisms' ability to produce more enzymes per m³ fermentation tank through genetic engineering



FERMENTATION

2. Optimizing industrial production

- Process optimization
- Equipment optimization
- Input optimization



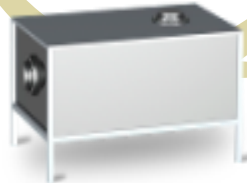
RAW MATERIALS



MICRO-ORGANISMS



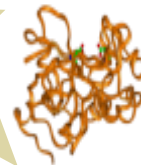
PURIFICATION



FORMULATION



MICROORGANISMS
TO BE INACTIVATED



ENZYMES



3. Improving the enzyme produced

Improving the efficacy of the enzymes through protein engineering to meet application conditions and process economy requirements

Project Overview



Project Participants



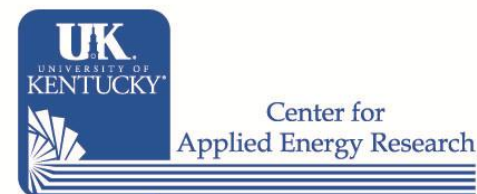
Ultrasonics & Aspen®



Full Process Analysis



Enzymes & Solvents



Kinetics & Bench-scale Tests

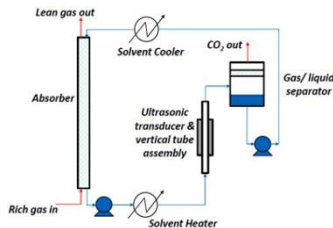
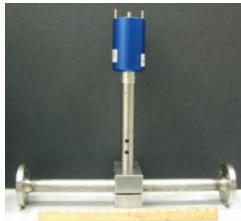
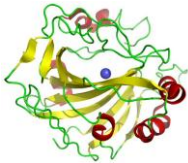
- DOE Project Manager: Andrew Jones
- Project Number: DE-FE0007741
- Total Project Budget: \$2,088,644
 - DOE: \$1,658,620
 - Cost Share: \$430,024
- Project Duration: Oct. 1, 2011 – March 31, 2015

DOE Program Objectives
 Develop solvent-based, post-combustion technology that

- Can achieve $\geq 90\%$ CO₂ removal from coal-fired power plants
- Demonstrates progress toward the DOE target of <35% increase in LCOE.

Project Objective

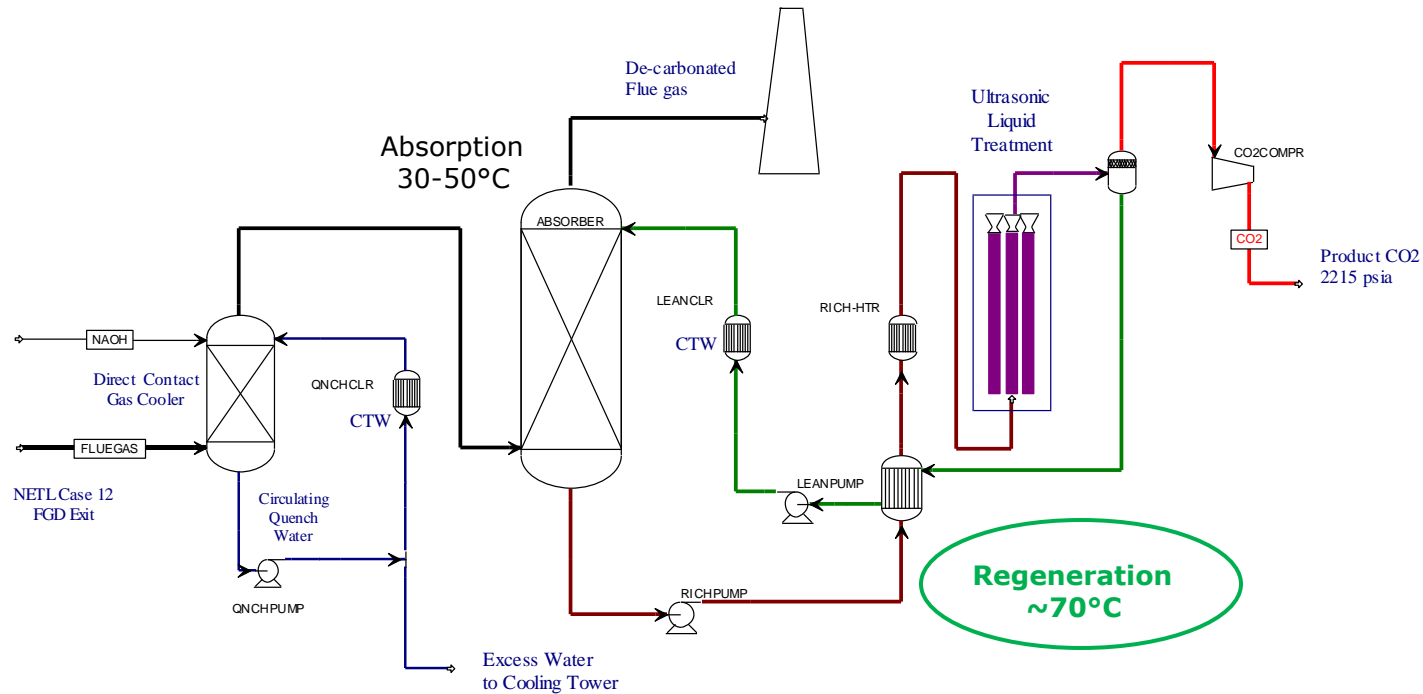
Complete a *bench-scale study* and corresponding full technology assessment to validate the potential in meeting the DOE Program Objectives of a *solvent-based post-combustion carbon dioxide capture* system that integrates



- a **low-enthalpy**, aqueous potassium carbonate-based solvent
- with an **absorption**-enhancing carbonic anhydrase enzyme catalyst
- and a flow through ultrasonic-enhanced **regenerator**
- in a **re-circulating** absorption-desorption process configuration

Challenges Encountered

Process Concept



Advantages

- Low enthalpy, benign solvent (catalyzed aq. 20% K₂CO₃)
 - K₂CO₃ ΔH_{rxn} 27 kJ/mol CO₂
 - MEA ΔH_{rxn} 83 kJ/mol CO₂
- Potential for ~50% regeneration energy vs. MEA

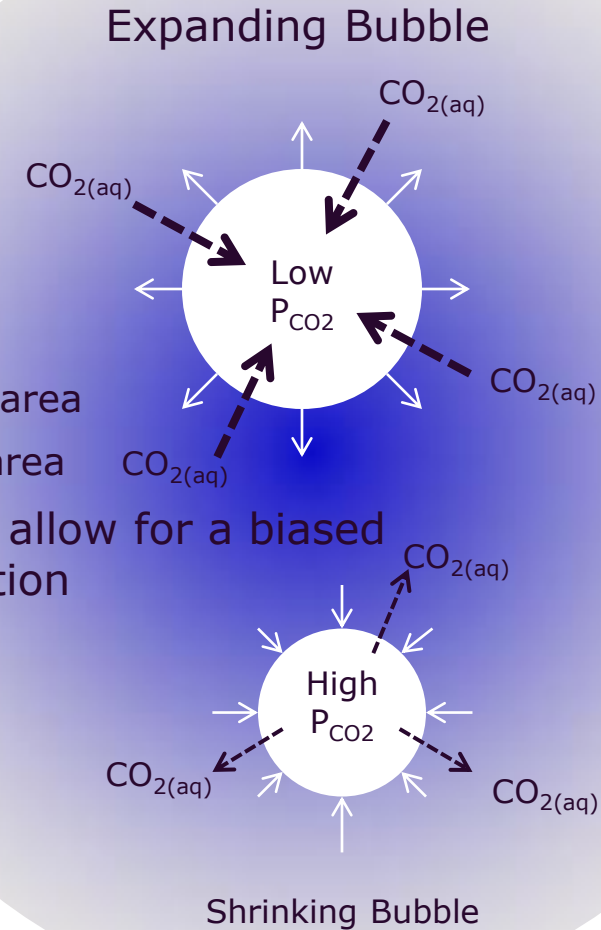
Challenges

- Demonstrate atmospheric regeneration at 70°C enabled by ultrasonics
- Demonstrate overall techno-economic feasibility
 - energy demand
 - enzyme requirement

Background on Ultrasonic Technology

- Rectified Diffusion Mechanism: [1]
 - Bubbles expand and shrink in an ultrasonic field
 - Expanding bubbles = lower pressure/ higher surface area
 - Shrinking bubbles = higher pressure/ lower surface area
 - Rectified diffusion results when expanding bubbles allow for a biased transfer of dissolved gas into the bubble from solution

- Proposed approach for solvent regeneration:
 - Create a population of seed bubbles
 - Grow the bubbles via rectified diffusion.
 - Frequency optimization likely required
 - Rapidly remove bubbles before they can dissolve



[1] Louisnard and Gomez (2003): *Theoretical predictions of rectified diffusion of air in water (1 bar, 26.5 kHz acoustic field)*

Project Schedule & Status Summary

- Task 1 – Project Management and Planning
- Task 2 – Process optimization
 - Batch-mode ultrasonics provided modest CO₂ release
 - Enzyme-solvent compatibility and absorption kinetics targets met
 - Integrated Bench-Scale system designed
- Task 3 – Initial Technical & Economic Feasibility
 - Indicated opportunity for 25% net efficiency improvement vs Case 10
- Task 4 – Bench Unit Procurement & Fabrication
 - Proto-type flow-through ultrasonic unit built & tested
 - Constructed bench-scale absorber and host rig with vacuum stripper
- Task 5 – Bench-scale Integration & Shakedown Testing
 - Commissioning and shakedown testing w/vacuum stripping in progress
 - Long-term enzyme stability and reclamation in progress
 - Initiating kinetics-based stripping simulation
- Task 6 – Bench-scale Testing
- Task 7 – Full Technology Assessment

Start
10/2011

CCTM
07/2012

BP2
01/2013

CCTM
07/2013

BP3
01/2014

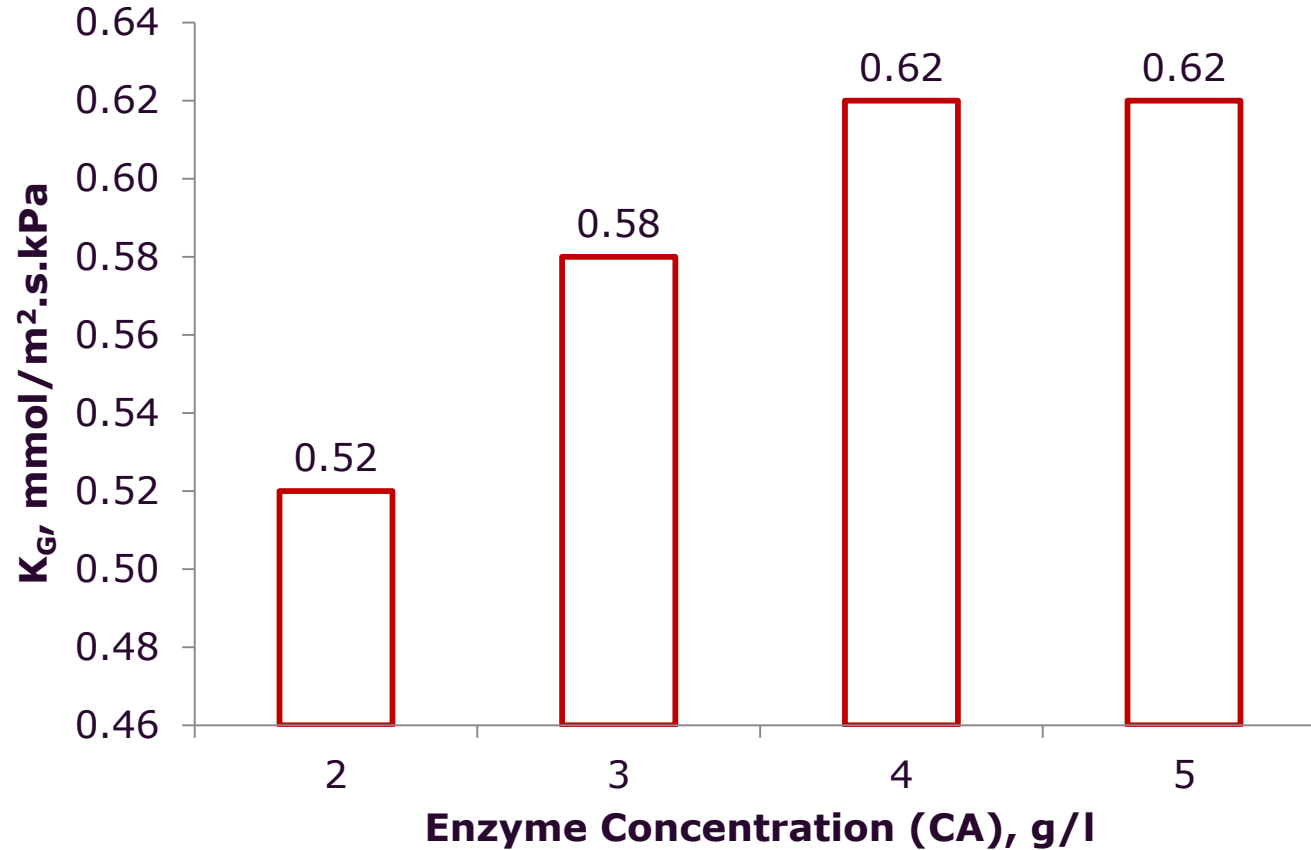
Preliminary Technical and Economic Feasibility

- Aspen Plus® (with Radfrac) used for Process modeling for absorption
- AspenTech's Capital Cost Estimator® along with budget supplier quotations used for Cost Estimation of the PCC Components
- Preliminary techno-economic evaluation for the process integrated with a subcritical coal-fired power plant was carried out indicating net efficiency improvement of up to 25% versus Case 10:

		Net efficiency, %	LCOE (\$/MWh _e)
	Case 10	24.9	119.6
Power Equivalent of 0.0911 kWh/lb of steam	Vacuum Regeneration	24.3	125.2
	Ultrasonic Regeneration	26.6	117.5
Power Equivalent of 0.0665 kWh/lb of steam	Vacuum Regeneration	30.0	112.9
	Ultrasonic Regeneration	31.4	108.9

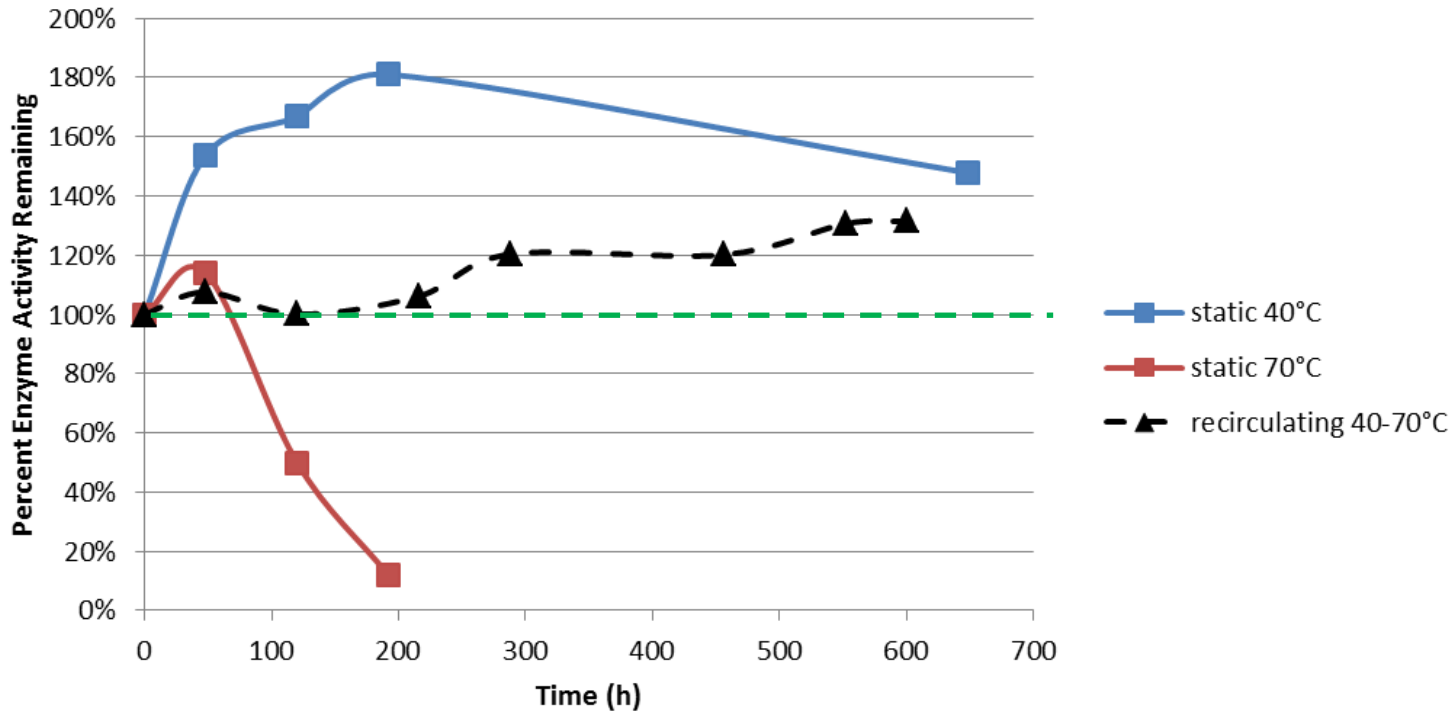
- Key underlying assumptions were:
 - Acceptable enhancement of CO₂ absorption rate via enzyme
 - Acceptable enzyme longevity in process
 - Ultrasonic regeneration in no more than two stages (1.5 kJe/ kg of solvent)
 - Vacuum regeneration at 6psia and 70°C

Acceptable CO₂ Absorption Rate



- Solvent: aq. 20% K₂CO₃ + carbonic anhydrase
- Demonstrated acceptable kinetics (mass transfer) with enzyme
- Temperature (30-50°C) had minimal impact

Acceptable (Lab Scale) Enzyme-solvent Longevity



Solvent: aq. 22% $K_2CO_3/KHCO_3$ with 3 g/L enzyme and adjusted to lean pH.

- Static incubations demonstrate high robustness at 40°C and limited robustness at 70°C.
- A more representative test (recirculating between 40-70°C) demonstrates high robustness across the temperature range.

Basis for Ultrasonic Regeneration Energy Projections

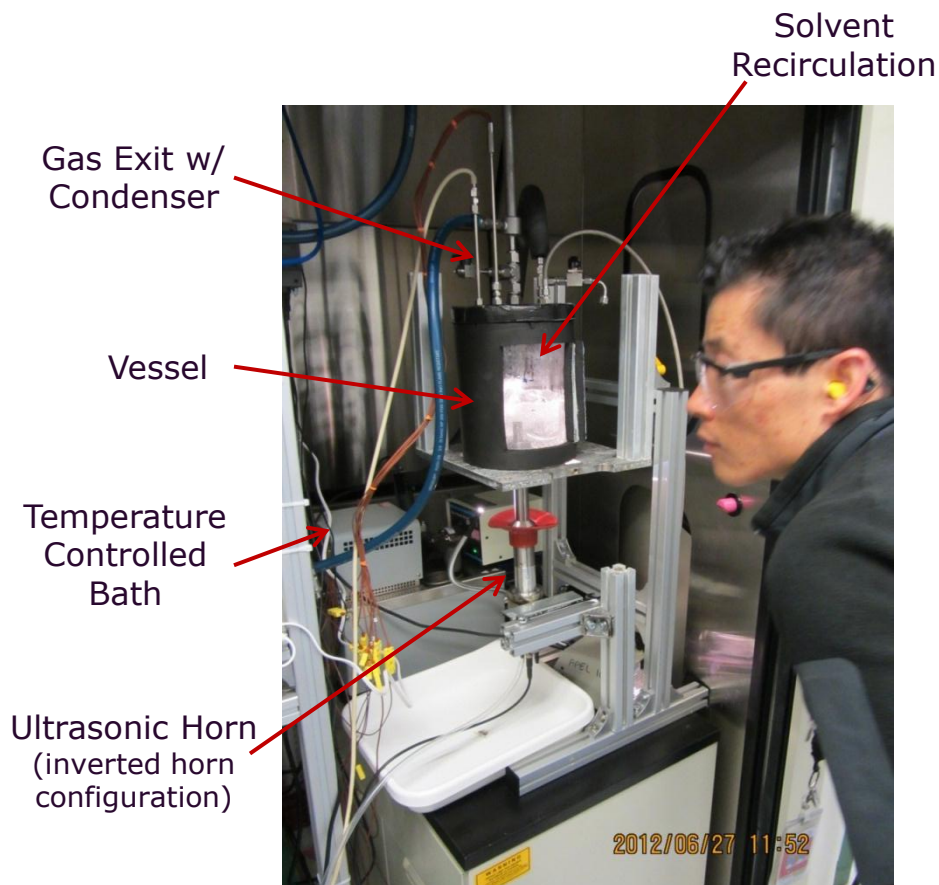
- ▶ Commercial water sterilization = 0.24 to 0.79 kJe/ kg of water
 - Based on developed applications for ship ballast treatment [2]

- ▶ Initial batch testing for CO₂ regeneration = 4.9 kJe/ kg of solvent
 - Laboratory horn used. Poor CO₂ removal (significant re-dissolution)
 - Demonstrated value = 10.3 kJe /mol of CO₂, 0.021 kg of CO₂ removal per kg of recirculated solvent recirculation assumed.

- ▶ Full-scale CO₂ regeneration system estimate = 1.5 kJe/ kg of solvent
 - Based on (conservative) tube sonication configuration
 - Equates to just over 11 MWe of parasitic power for the ultrasonic system in the 500 MWe reference system)

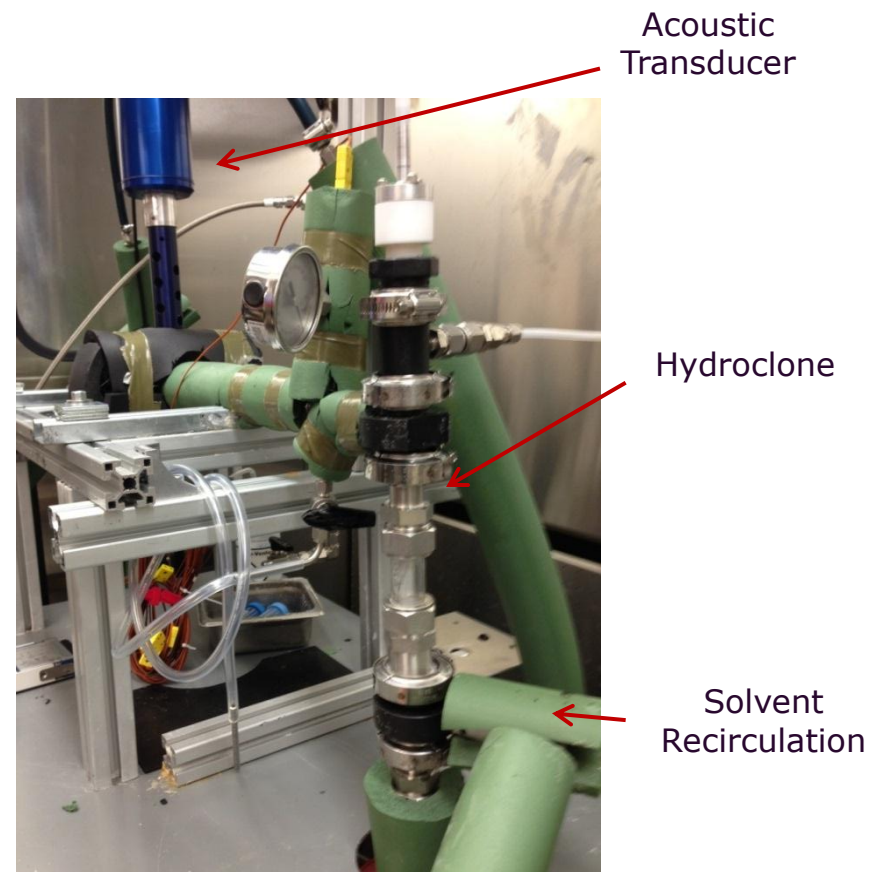
[2] "Ballast water treatment technology, Current status," February 2010
(http://www.lr.org/Images/BWT0210_tcm155-175072.pdf)

Ultrasonic Testing Platforms



Batch System

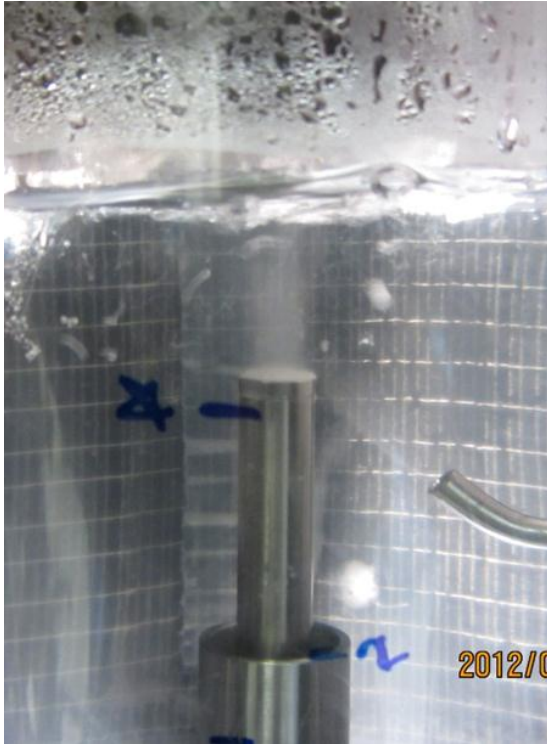
Can introduce ultrasonic power while maintaining temperature to within 2°C.



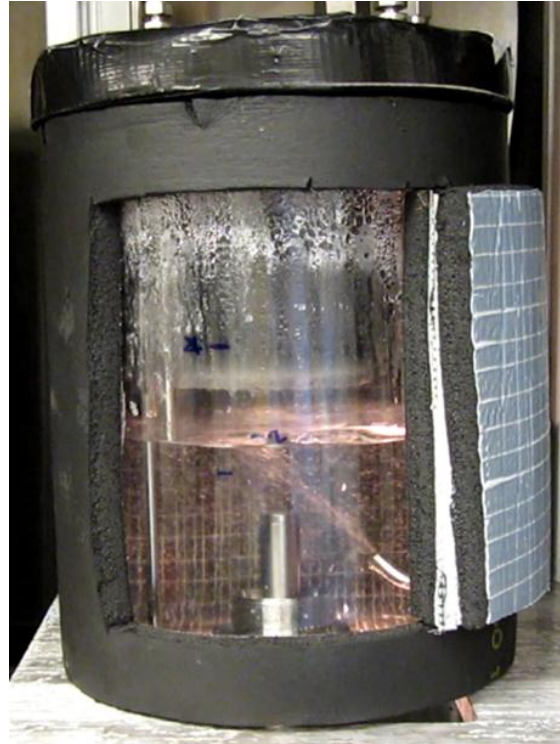
Semi-Continuous System

*Large reservoir of solvent recirculated.
 Gas separated after sonication via
 hydroclone*

Initial Batch Ultrasonic Experiments



*Pure Water at 70°C
– With Sonication*



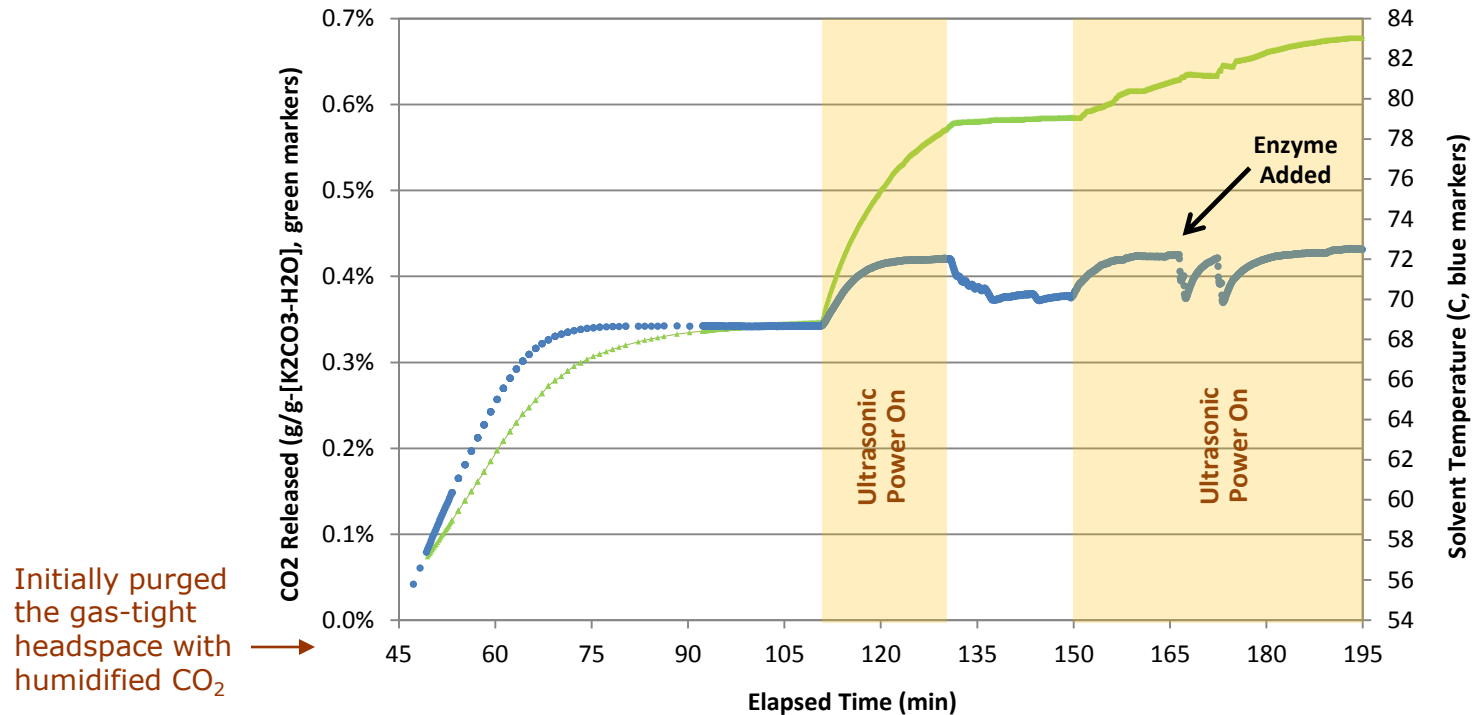
*Loaded Solvent at 70°C
– No Sonication*



*Loaded Solvent at 70°C
– With Sonication*

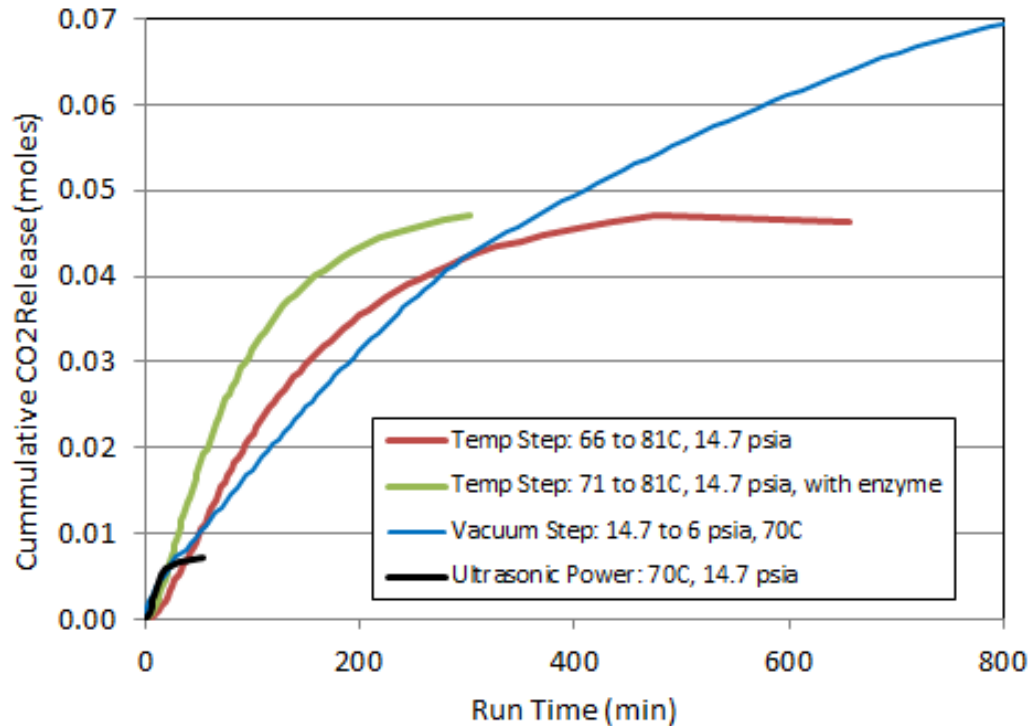
- Significant agitation/ bubbling observed when ultrasonic power added to loaded K_2CO_3 solution at 70°C

Batch Test Results for Ultrasonic Regeneration



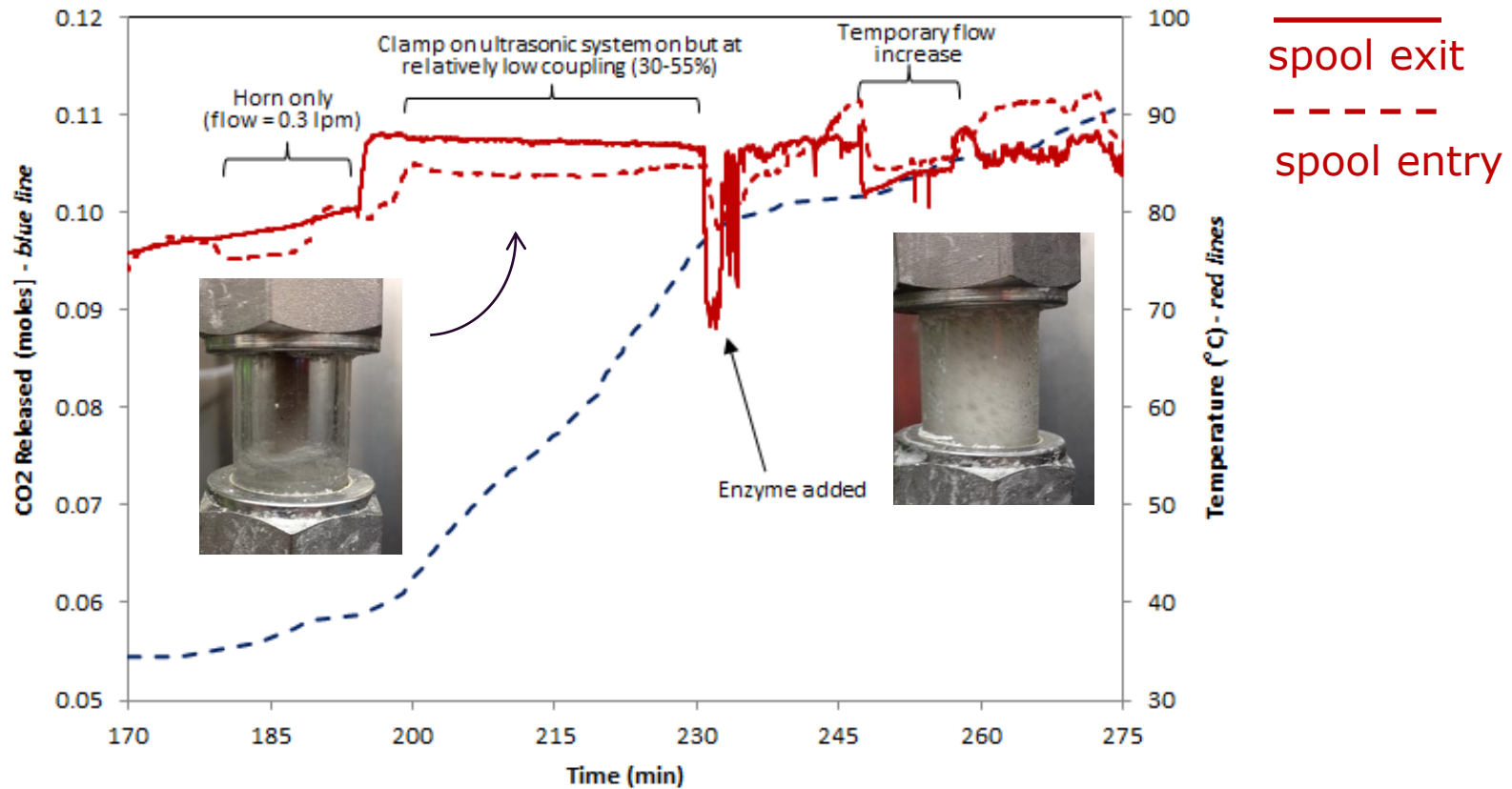
- ▶ Testing with 20 wt% K₂CO₃ solvent loaded to 4.6 wt% CO₂
- ▶ ASPEN (equilibrium) projections of CO₂ release at 6 psia = 0.96%
- ▶ Total CO₂ release observed = 0.67% (0.25% from ultrasonic effect) – likely impacted by re-dissolution of CO₂
- ▶ Slow CO₂ release rates observed – also likely impacted by re-dissolution of CO₂

Comparison of Batch-mode Regeneration



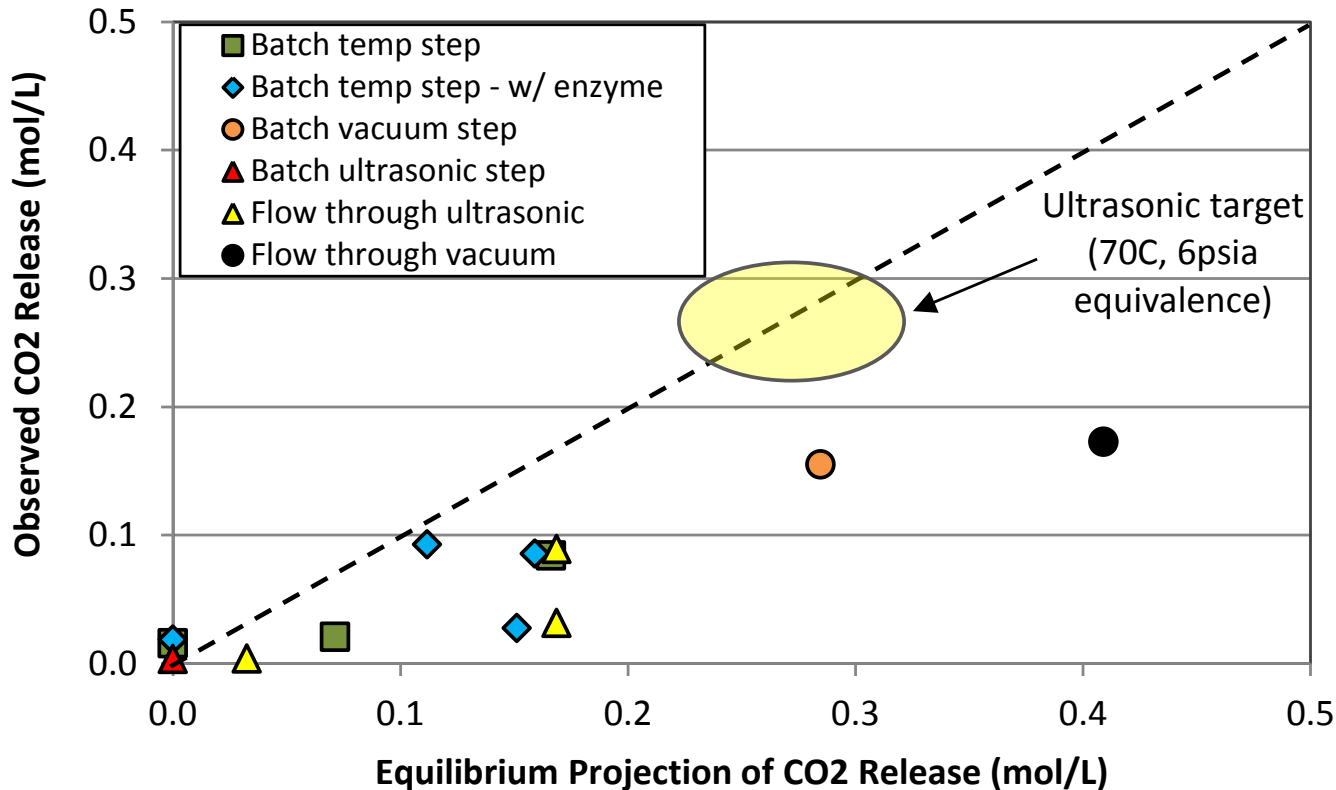
- ▶ All tests with 20 wt% K₂CO₃ – temp tests at ~82% (converted to bicarbonate), vacuum and ultrasonic tests at 72%
- ▶ Similar kinetic rates (initial part of curves) but higher with enzyme – kinetic limitation?
- ▶ Total CO₂ release low for ultrasonic test – CO₂ re-dissolution suspected

Ultrasonic Flow-Through Results



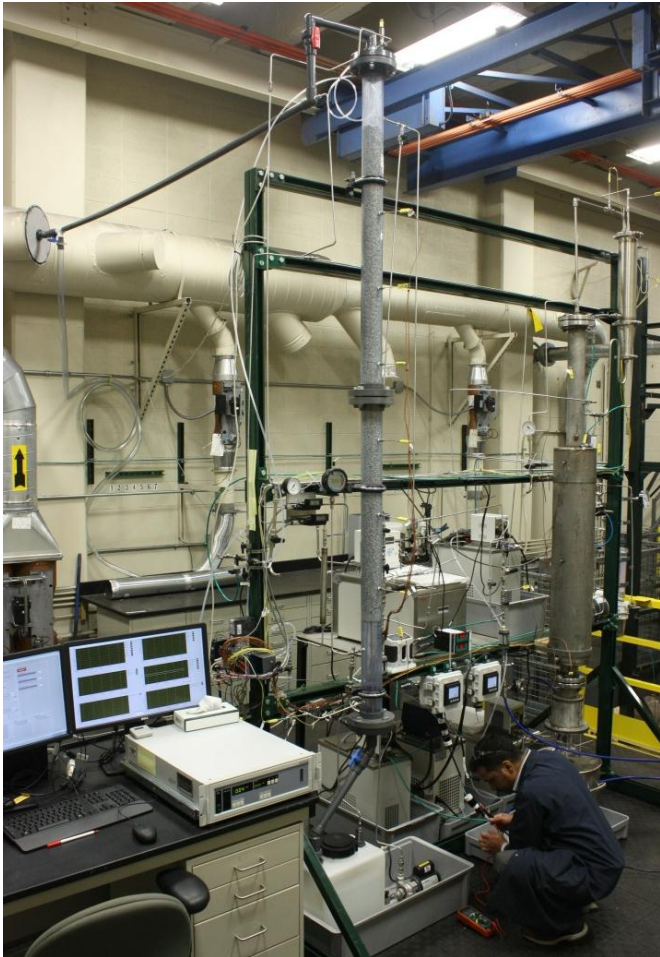
- CO₂ release rate similar to batch studies – can be explained by temperature increase alone
- Enzyme additions unexpectedly decreased release rate – likely due to foaming

Summary of Regeneration Testing Results



- Multiple passes (5+) required for significant CO₂ release from both vacuum and ultrasonic flow through tests – kinetic limitation suspected
- Ultrasonic flow through results within temperature-driven projections; not in line with 70°C, 6psia vacuum target
- Current ultrasonics configuration delivers insufficient CO₂ release

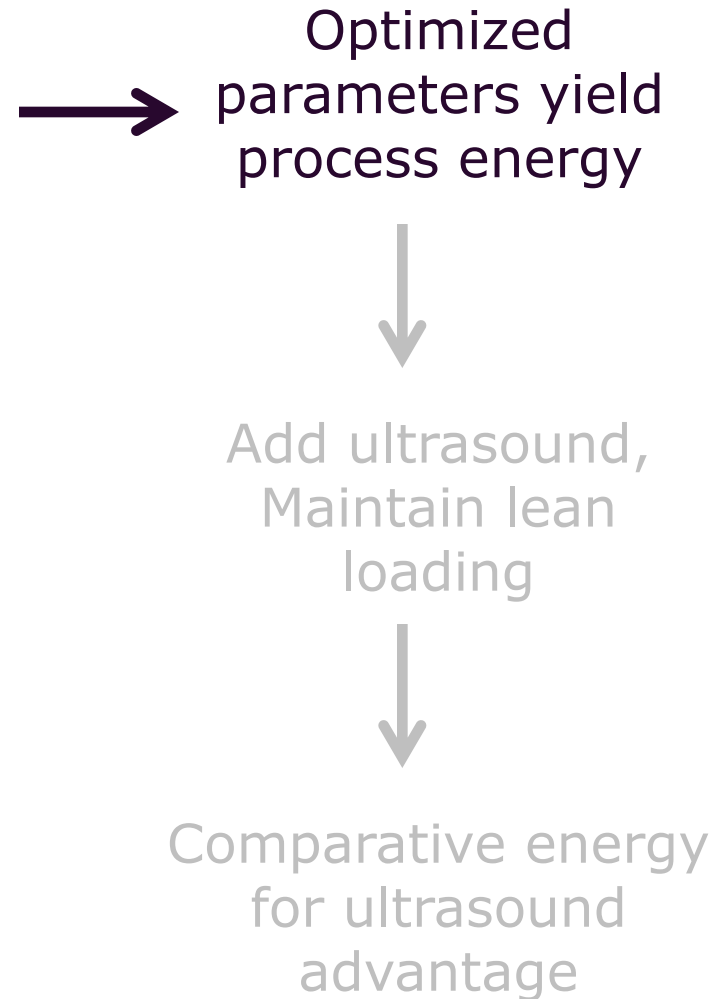
Bench-scale Demonstration Unit Status



- Design capabilities:
 - Dual regeneration sources (vacuum and ultrasonic)
 - Able to assess long-term enzyme stability
 - Able to assess mass transfer
- Construction complete:
 - Host rig framework
 - Absorber
 - Vacuum regeneration
 - Heat transfer
 - Instrumentation check and calibrations
- Unit commissioning for vacuum process in progress

Key Bench-scale Operational Parameters

- Flow rates
 - Gas: 10- 30 SLPM
 - Liquid : 100-300 ml/min
- Liquid temperature
 - Absorber inlet: 30-40 °C
 - Stripper outlet: 70-80 °C
- Stripper pressure: 0.25-0.4 atm
- Enzyme dose: 3-5 g/L



Conclusions and Next Steps

- Target absorption kinetics and enzyme robustness measured
- Visual evidence of ultrasonic effect shown in batch system
- Preliminary techno-economic evaluation indicated potential for net efficiency improvement of up to 25% versus Case 10
- Construction of bench-scale absorption column with vacuum regeneration completed and commissioning in progress
- Flow-through bench-scale ultrasonic regeneration system was assembled and tested
 - CO₂ release rates below single-pass stripping target for the project
 - Low CO₂ release rates may point toward a kinetic limitation in stripping; enzyme catalyst could help overcome this limitation
 - Ultrasonics in current configuration delivers insufficient CO₂ release
- Project now focuses on validating the potential for low temperature regeneration by developing a rate-based simulation for vacuum stripping corroborated by data from bench-scale testing



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